



# Colloque Annuel du GdR CNRS HydroGEMM

Mardi 05 novembre 2024

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## Comprehensive Overview from Earth cycle to experimental studies

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**GDR**  
H y d r o G E M M

Hydrogemm 2024  
Raphaël Josse

Natural Hydrogen

**What if everything was hidden in its **Earth cycle** ?**

# About me ...

**2023**

## **Internship with Isabelle Moretti (ISTeP)**

- Remote sensing
- Sources rock for natural hydrogen generation

**2024**

**Graduated in MS.C Geology** from Sorbonne University, France

## **Internship with Ema Frery (CSIRO)**

- Hydrogen cycle
- Long-term monitoring
- Artefacts measured on field

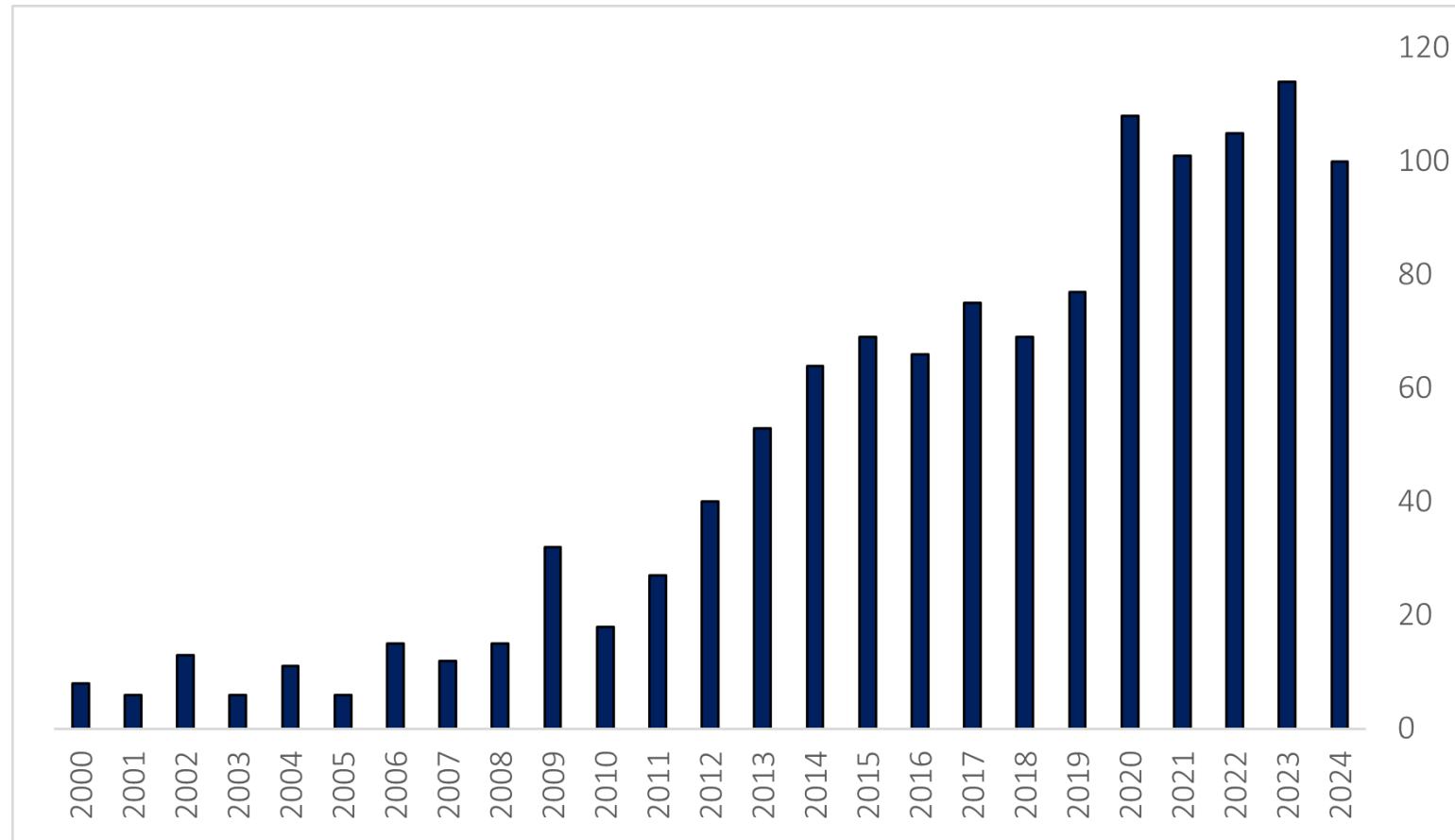
**Now**

## **Research Geologist Engineer at IFPEN**



- Hydrogen exploration strategies
- Gas potentiel through expermentals studies

# Scientific papers related to Natural hydrogen in Geology



Source : WebOfScience, « natural hydrogen geology from 2000 to 2024 »

**Scientifics papers related to  
Natural hydrogen **Cycle** in Geology ?**

**And the answer is ...**

**Almost none**

# Need to create a more accurate H<sub>2</sub> cycle

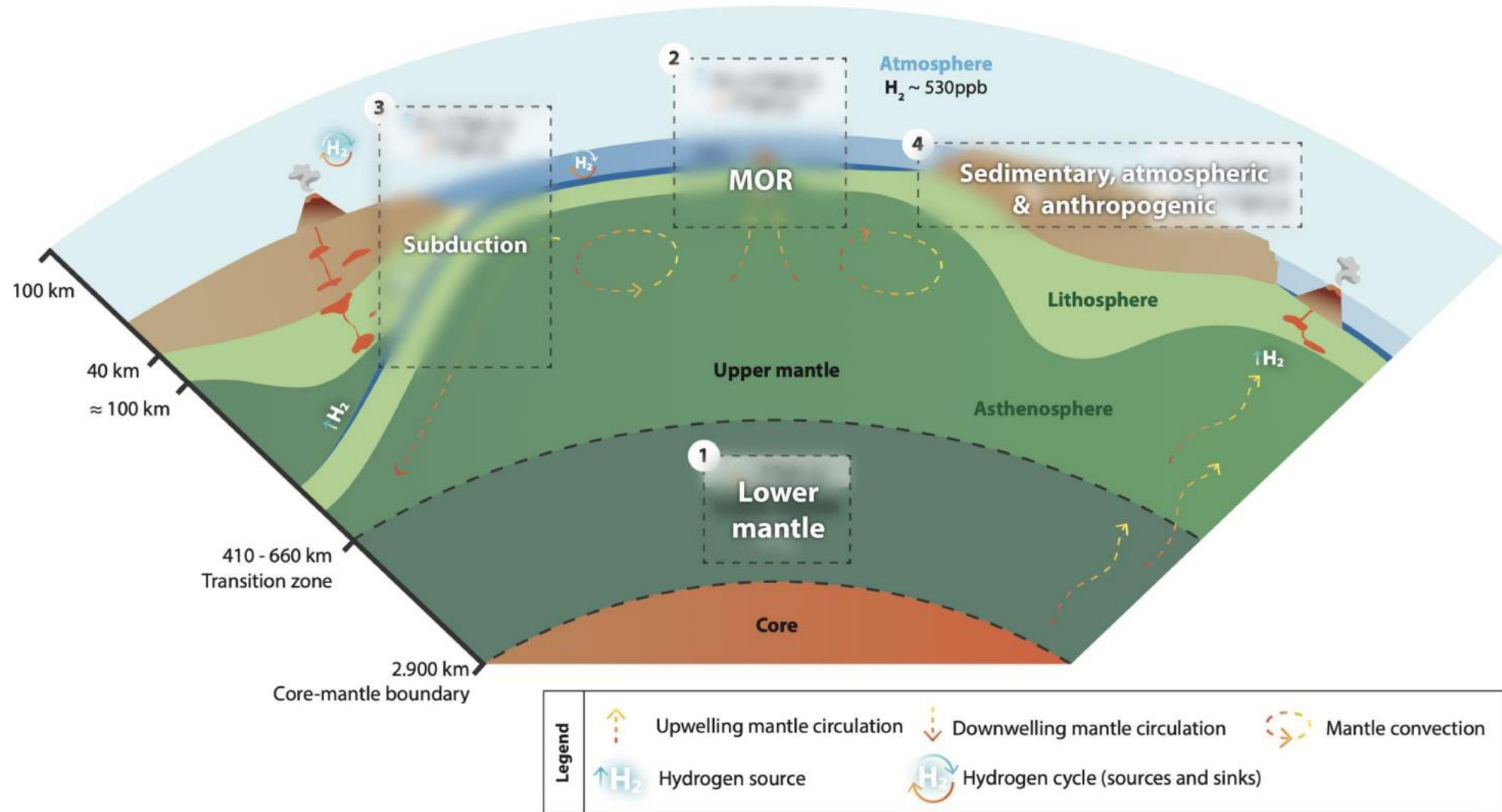
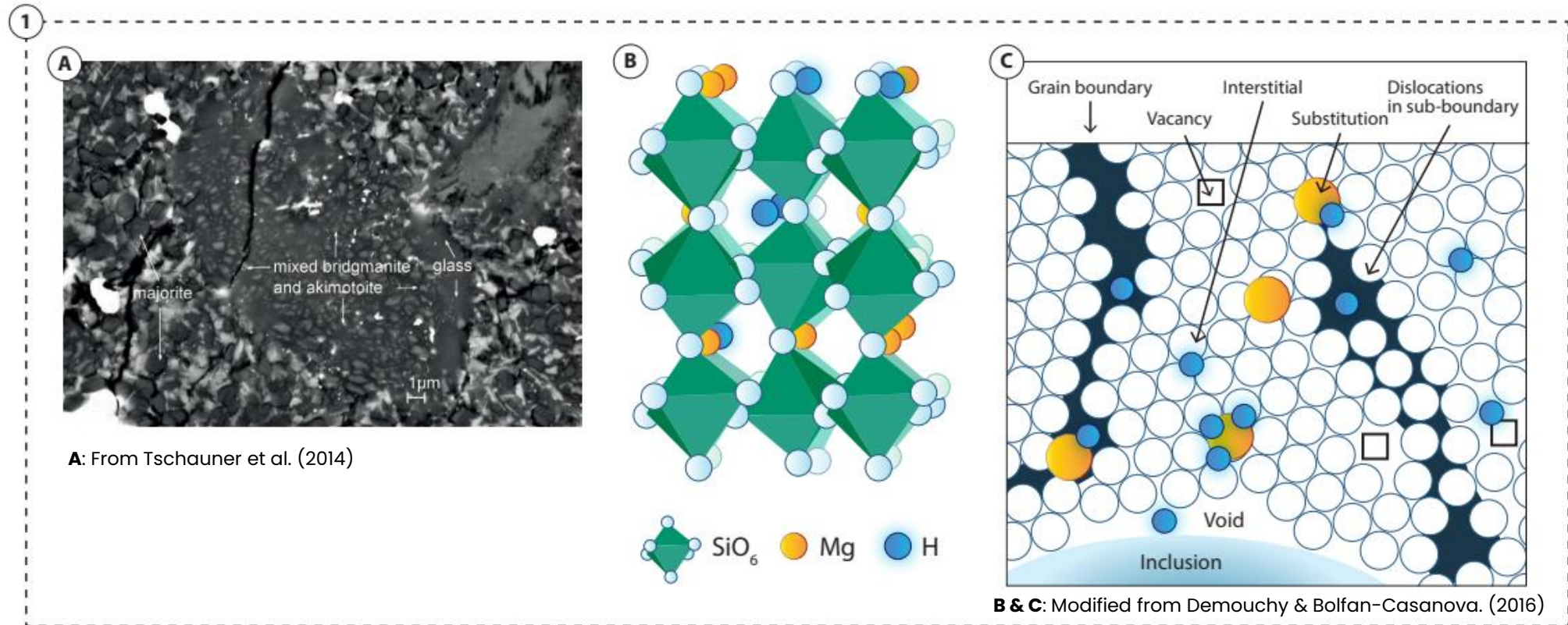


Figure 1: Main sketch of H<sub>2</sub> cycle on Earth

# 1 - H<sub>2</sub> in the Lower mantle



**A** - SEM image of  
bridgmanite



**B** - 3D view of crystal  
structure  $\text{MgSiO}_3$   
Perovskite

**C** - 2D view of crystal  
structure  $\text{MgSiO}_3$   
Perovskite

Figure 2: H<sub>2</sub> reservoirs in the lower mantle,  
especially in a Mg-perovskite structure.



# 2 - H<sub>2</sub> in the MOR

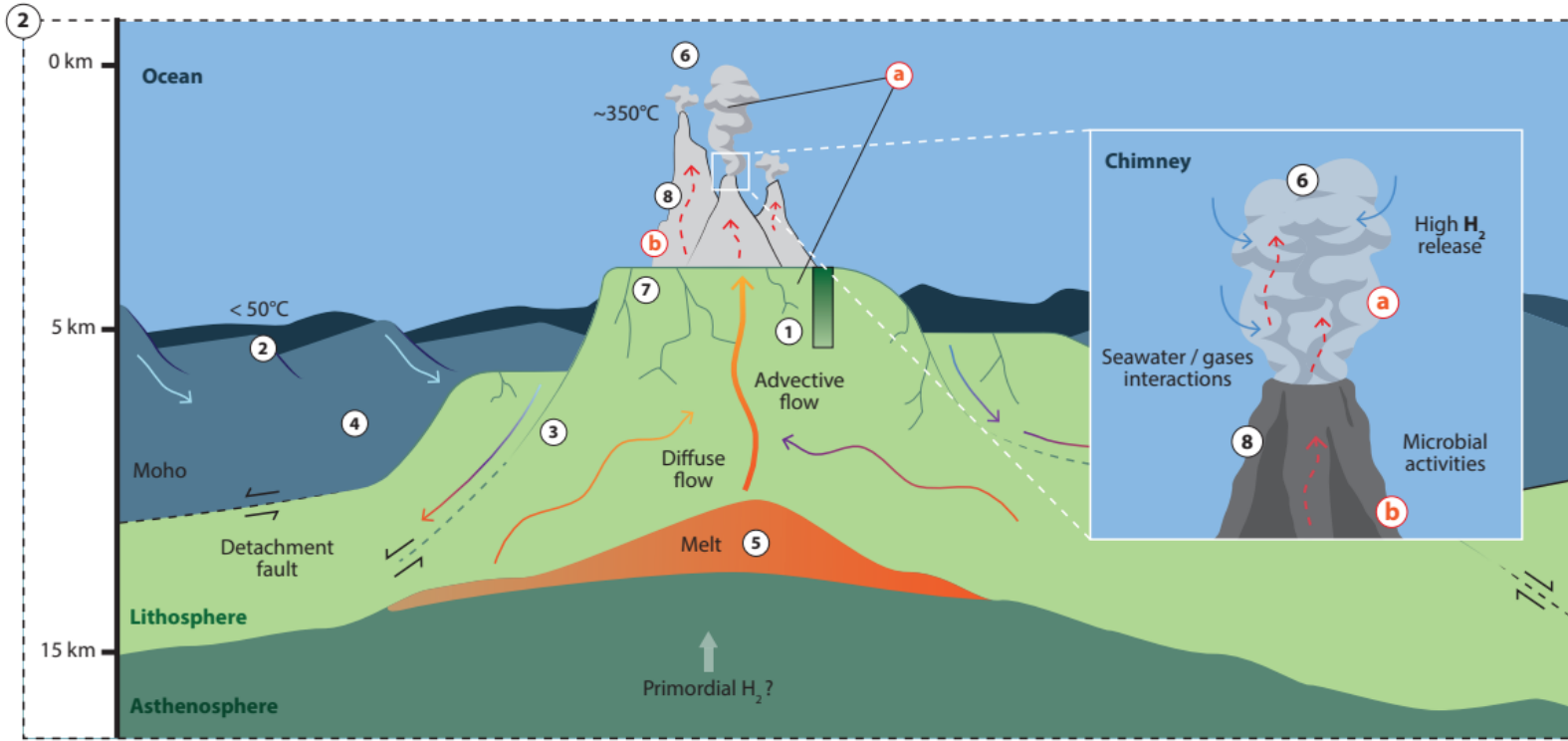


Figure 3: H<sub>2</sub> cycle associated with Mid-oceanic ridges in hydrothermal vents

Sources	
<b>Abiotic</b>	
① Serpentinization $(Mg, Fe)_2SiO_4 + H_2O \rightarrow Mg_3Si_2O_5(OH)_4 + Mg(OH)_2 + Fe_3O_4 + H_2$	③ Rock fracturing $Si \cdot + H_2O \rightarrow SiOH + H \cdot$ $H \cdot + H \cdot \rightarrow H_2$
② Oceanic crust weathering $2FeO + 4H_2O \rightarrow 2Fe(OH)_3 + H_2$	④ Water radiolysis $H_2O + (\alpha, \beta, \gamma) \rightarrow H \cdot + OH \cdot + H^+ + OH^- + e^-$ $e^-_{(aq)} + H^+ \rightarrow H \cdot$ $H \cdot + H \cdot \rightarrow H_2$
⑤ Magma crystallization $3FeO_{(magma)} + H_2O_{(magma)} \rightarrow (FeO \cdot Fe_2O_3)_{(rock)} + H_2$	⑦ Pyritization and sulphur interactions $FeS + H_2S \rightarrow FeS_2 + H_2$ $H_2S + 2H_2O \leftrightarrow SO_2 + 3H_2$
⑥ Magma degassing $CO + H_2O \leftrightarrow CO_2 + H_2$ $SO_2 + 2H_2O \leftrightarrow H_2SO_4 + H_2$ $H_2S + 2H_2O \leftrightarrow SO_2 + 3H_2$ $CH_4 + 2H_2O \leftrightarrow CO_2 + 4H_2$	⑧ Nitrogen fixation $N_2 + 8H^+ + 16ATP + 8e^- \rightarrow 2NH_3 + 16ADP + H_2$
<b>Biotic</b>	

Sinks	Legend
<b>Abiotic</b>	
(a) <b>Aerobic</b> Hydrogen oxidation $2H_2 + 2O_2 \rightarrow 2H_2O$ $5H_2 + 2NO_3^- \rightarrow N_2 + 4H_2O + 2OH^-$	<b>Fluid / rock interactions</b> → Cold seawater infiltration → High fluid modification → Hot hydrothermal fluid rise - - - Hydrothermal plume ■ Basalts / Lavas ■ Gabbroic rocks ■ Peridotite + Serpentinization gradient -
<b>Anaerobic</b> Sulfate reduction $4H_2 + 2H^+ + SO_4^{2-} \rightarrow H_2S + 4H_2O$	
Nitrate reduction $H_2 + NO_3^- \rightarrow NO_2^- + H_2O$	
Carbone reduction $4H_2 + CO_2 \rightarrow CH_4 + 2H_2O$	
<b>Biotic</b>	
(b) Microbial consumption $2H^+ + 2e^- \leftrightarrow H_2$	

# 3 - H<sub>2</sub> in Subduction

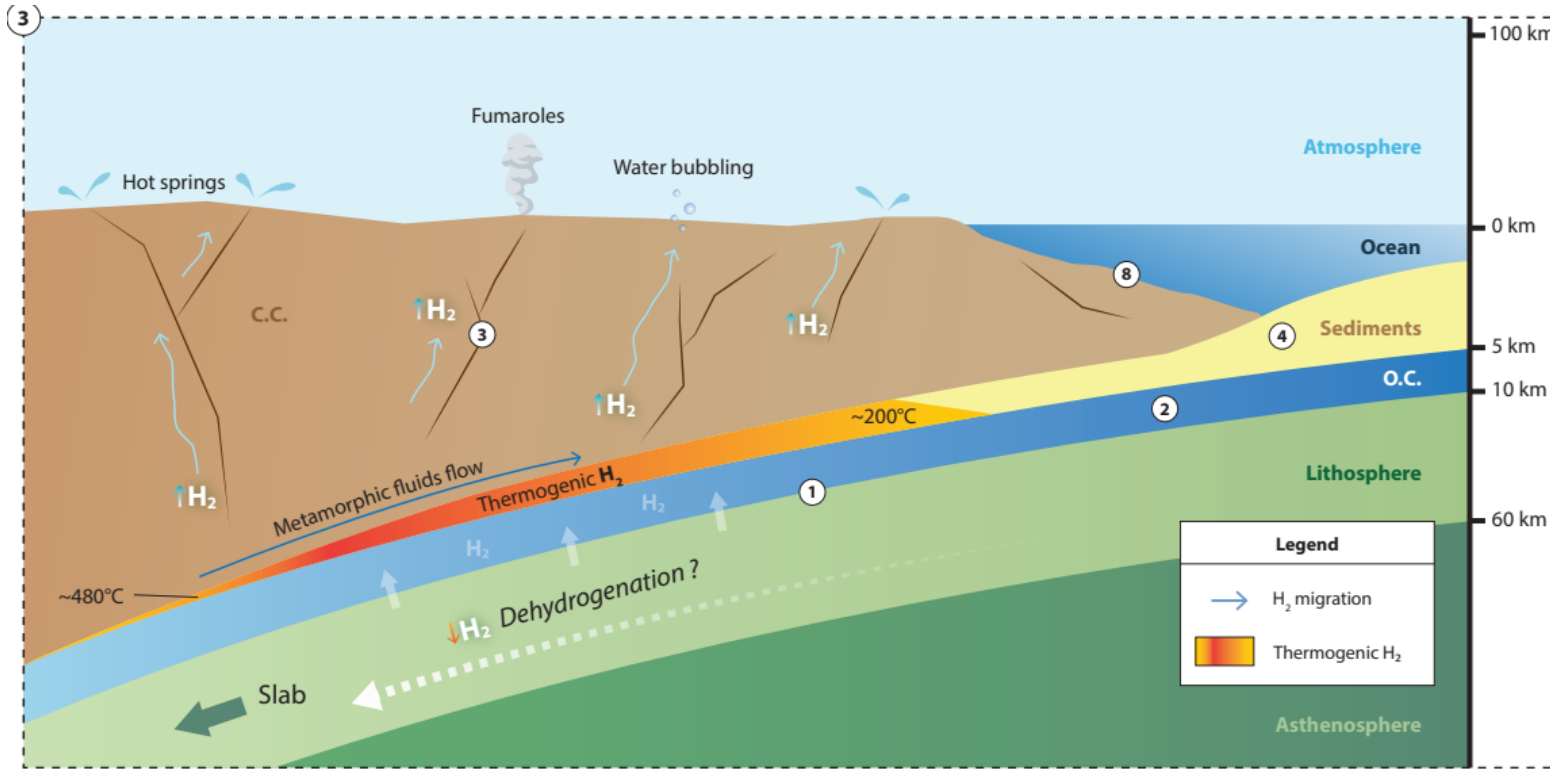


Figure 4: H<sub>2</sub> cycle associated with Subduction

Sources	
<b>Abiotic</b>	
① Serpentinization $(\text{Mg,Fe})_2\text{SiO}_4 + \text{H}_2\text{O} \rightarrow \text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4 + \text{Mg}(\text{OH})_2 + \text{Fe}_3\text{O}_4 + \text{H}_2$	③ Rock fracturing $\text{Si} \cdot + \text{H}_2\text{O} \rightarrow \text{SiOH} + \text{H} \cdot$ $\text{H} \cdot + \text{H} \cdot \rightarrow \text{H}_2$
② Oceanic crust weathering $2\text{FeO} + 4\text{H}_2\text{O} \rightarrow 2\text{Fe}(\text{OH})_3 + \text{H}_2$	④ Water radiolysis $\text{H}_2\text{O} + (\alpha, \beta, \gamma) \rightarrow \text{H} \cdot + \text{OH} \cdot + \text{H}^+ + \text{OH}^- + \text{e}^-$ $\text{e}^-_{(\text{aq})} + \text{H}^+ \rightarrow \text{H} \cdot$ $\text{H} \cdot + \text{H} \cdot \rightarrow \text{H}_2$
⑤ Magma crystallization $3\text{FeO}_{(\text{magma})} + \text{H}_2\text{O}_{(\text{magma})} \rightarrow (\text{FeO} \cdot \text{Fe}_2\text{O}_3)_{(\text{rock})} + \text{H}_2$	⑥ Magma degassing $\text{CO} + \text{H}_2\text{O} \leftrightarrow \text{CO}_2 + \text{H}_2$ $\text{SO}_2 + 2\text{H}_2\text{O} \leftrightarrow \text{H}_2\text{SO}_4 + \text{H}_2$ $\text{H}_2\text{S} + 2\text{H}_2\text{O} \leftrightarrow \text{SO}_2 + 3\text{H}_2$ $\text{CH}_4 + 2\text{H}_2\text{O} \leftrightarrow \text{CO}_2 + 4\text{H}_2$
⑦ Pyritization and sulphur interactions $\text{FeS} + \text{H}_2\text{S} \rightarrow \text{FeS}_2 + \text{H}_2$ $\text{H}_2\text{S} + 2\text{H}_2\text{O} \leftrightarrow \text{SO}_2 + 3\text{H}_2$	⑧ Nitrogen fixation $\text{N}_2 + 8\text{H}^+ + 16\text{ATP} + 8\text{e}^- \rightarrow 2\text{NH}_3 + 16\text{ADP} + \text{H}_2$
<b>Biotic</b>	
<b>Sinks</b>	
<b>Abiotic</b>	
a) Aerobic Hydrogen oxidation $2\text{H}_2 + 2\text{O}_2 \rightarrow 2\text{H}_2\text{O}$ $5\text{H}_2 + 2\text{NO}_3^- \rightarrow \text{N}_2 + 4\text{H}_2\text{O} + 2\text{OH}^-$	
Anaerobic Sulfate reduction $4\text{H}_2 + 2\text{H}^+ + \text{SO}_4^{2-} \rightarrow \text{H}_2\text{S} + 4\text{H}_2\text{O}$	
Nitrate reduction $\text{H}_2 + \text{NO}_3^- \rightarrow \text{NO}_2^- + \text{H}_2\text{O}$	
Carbone reduction $4\text{H}_2 + \text{CO}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O}$	
<b>Biotic</b>	
b) Microbial consumption $2\text{H}^+ + 2\text{e}^- \leftrightarrow \text{H}_2$	
<b>Legend</b>	
<i>Fluid / rock interactions</i>	
→ Cold seawater infiltration	
→ High fluid modification	
→ Hot hydrothermal fluid rise	
- - - Hydrothermal plume	
■ Basalts / Lavas	
■ Gabbroic rocks	
■ Peridotite	
+ Serpentinization gradient	

# 4 - H<sub>2</sub> in Sedimentary, Atmospheric and Anthropogenic

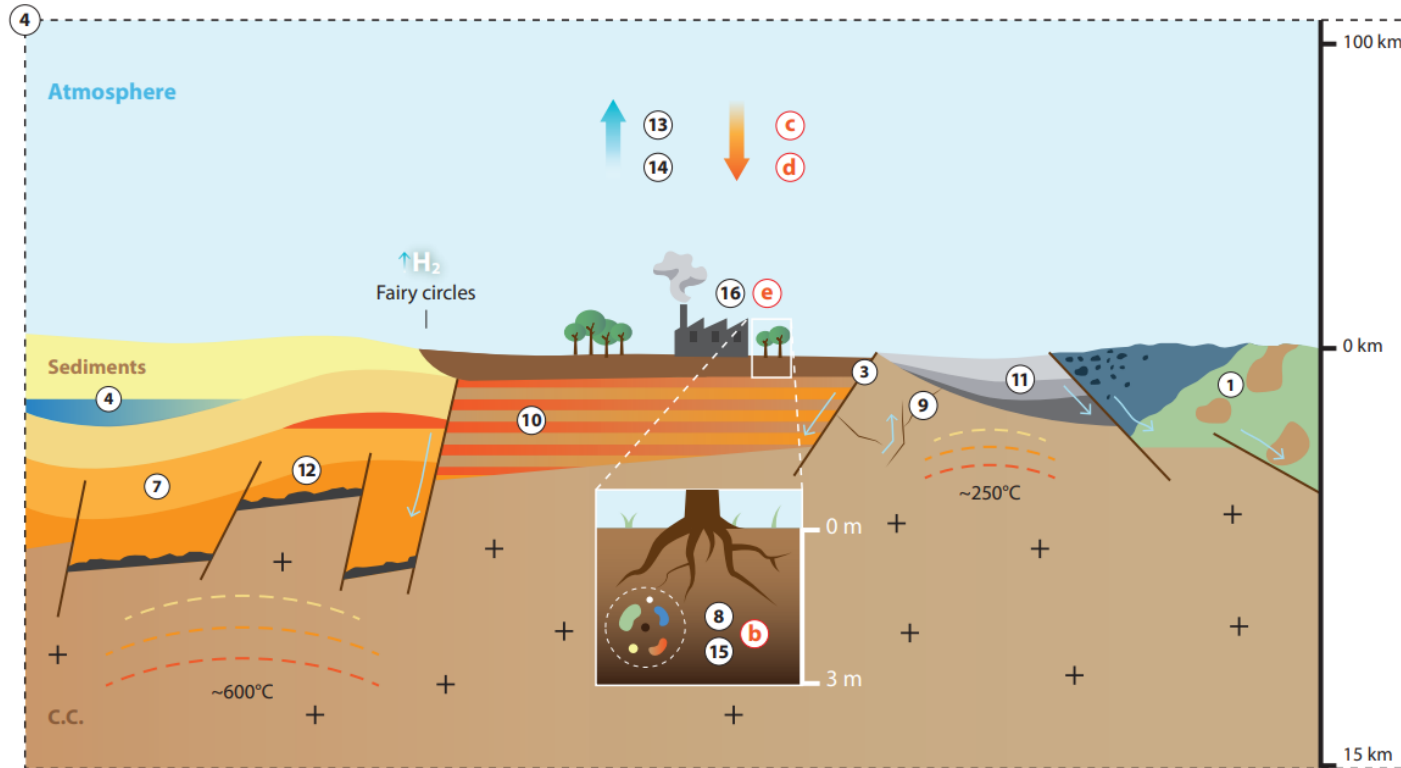


Figure 5: H<sub>2</sub> cycle associated with sedimentary, atmospheric and anthropogenic contexts.

Sources	
<b>Abiotic</b>	
① Serpentinization $(\text{Mg,Fe})_2\text{SiO}_4 + \text{H}_2\text{O} \rightarrow \text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4 + \text{Mg}(\text{OH})_2 + \text{Fe}_3\text{O}_4 + \text{H}_2$	④ Water radiolysis $\text{H}_2\text{O} + (\alpha, \beta, \gamma) \rightarrow \text{H}\cdot + \text{OH}\cdot + \text{H}^+ + \text{OH}^- + \text{e}^-$ $\text{e}^- + \text{H}^+ \rightarrow \text{H}\cdot$ $\text{H}\cdot + \text{H}\cdot \rightarrow \text{H}_2$
③ Rock fracturing $\text{Si}\cdot + \text{H}_2\text{O} \rightarrow \text{SiOH} + \text{H}\cdot$ $\text{H}\cdot + \text{H}\cdot \rightarrow \text{H}_2$	⑦ Pyritization and sulphur interactions $\text{FeS} + \text{H}_2\text{S} \rightarrow \text{FeS}_2 + \text{H}_2$ $\text{H}_2\text{S} + 2\text{H}_2\text{O} \leftrightarrow \text{SO}_2 + 3\text{H}_2$
⑨ Hydration of biotite $\text{KFe}_3^{2+}(\text{AlSi}_3)\text{O}_{10}(\text{OH})_2 + 2\text{H}_2\text{O} \rightarrow \text{Fe}_2^{3+}\text{O}_3 + \text{Fe}_3\text{O}(\text{OH}) + \text{H}_2$	⑪ Siderite weathering $3\text{FeCO}_3 + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + 3\text{CO}_2 + \text{H}_2$
⑩ BIF weathering $2\text{Fe}^{2+}\text{O} + \text{H}_2\text{O} \rightarrow \text{Fe}^{3+}2\text{O}_3 + \text{H}_2$	⑫ Pyrolysis of shales/coals
<b>Atmospheric</b>	
⑬ CH <sub>4</sub> photochemical oxidation $\text{CH}_4 + \text{OH} \rightarrow \text{CH}_3 + \text{H}_2\text{O}$	⑭ VOC photochemical oxidation $\text{HCHO} + \text{h}\nu \rightarrow \text{H}_2 + \text{CO}$
<b>Biotic</b>	
⑧ Nitrogen fixation $\text{N}_2 + 8\text{H}^+ + 16\text{ATP} + 8\text{e}^- \rightarrow 2\text{NH}_3 + 16\text{ADP} + \text{H}_2$	⑮ Biological production
<b>Anthropogenic</b>	
⑯ Anthropogenic hydrogen emissions	

Sinks	Legend
<b>Abiotic</b>	
① OH oxidation $\text{H}_2 + \text{OH} \rightarrow \text{H}_2\text{O} + \text{H}\cdot$	Water infiltration
② Planetary air leaks	Geothermal gradient
<b>Biotic</b>	
③ Microbial consumption $2\text{H}^+ + 2\text{e}^- \leftrightarrow \text{H}_2$	Microbial life
<b>Anthropogenic</b>	
④ FTT reaction $n\text{CO} + (2n + 1)\text{H}_2 \rightarrow \text{C}_n\text{H}_{2n+2} + n\text{H}_2\text{O}$	Banded-Iron Formations (BIF)
	Siderite rocks
	<b>Ophiolite complex</b>
	Basalts / Lavas
	Gabbroic rocks
	Chromite
	Peridotite

# Conclusion

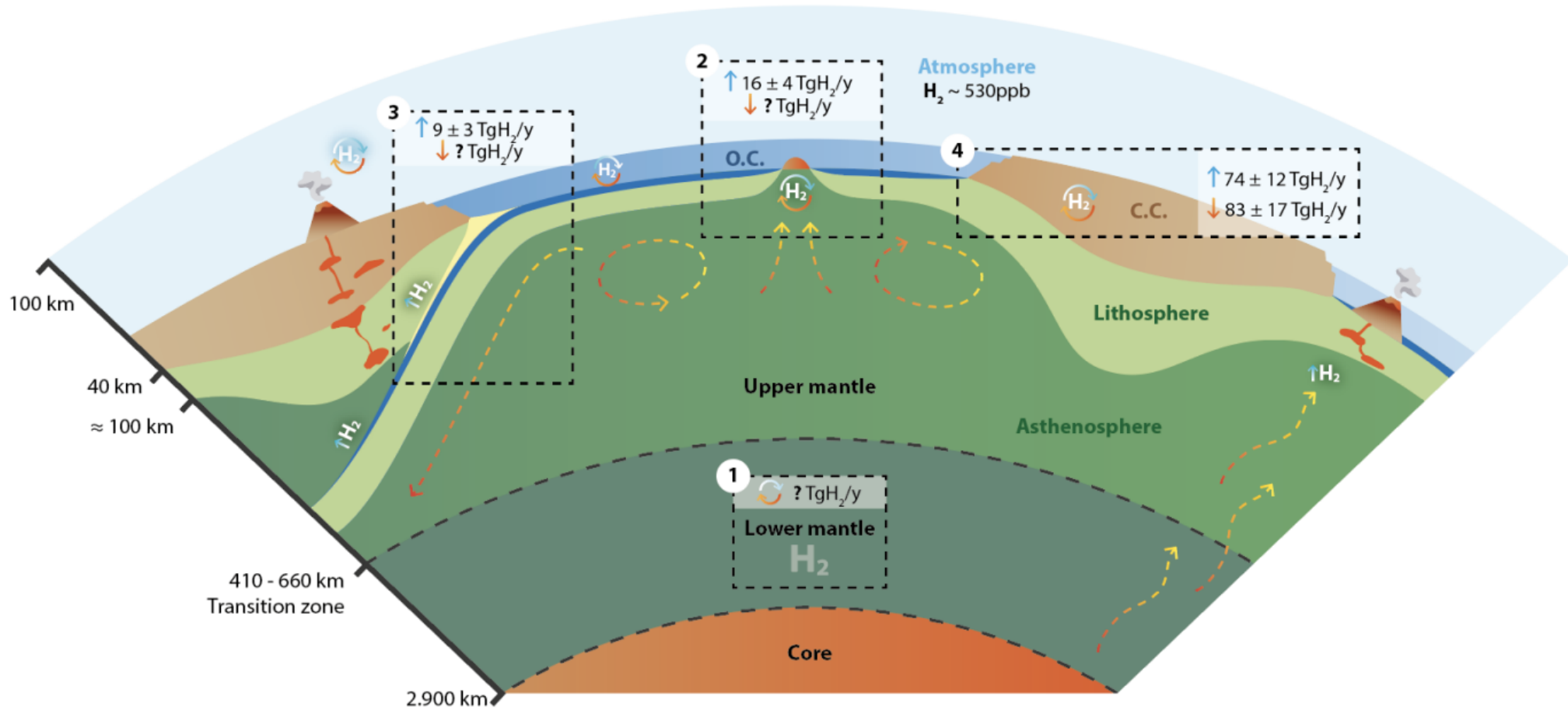


Figure 6: Final H<sub>2</sub> cycle with total flux

## Gaps:

- Volume on each compartment
- Interactions between each domain
- Processes overlooked

What's **next** ?

# **Create and Calibrate H<sub>2</sub> sensors**

for hydrogen exploration

# Let's play with experiments with H<sub>2</sub> injections

- **1<sup>st</sup> step:** Use right sensors

Semiconductor Gas Sensors (**SGX**)

- Silicon layers

Metal Oxide Sensors (**MOX**)

- Zinc or copper layers

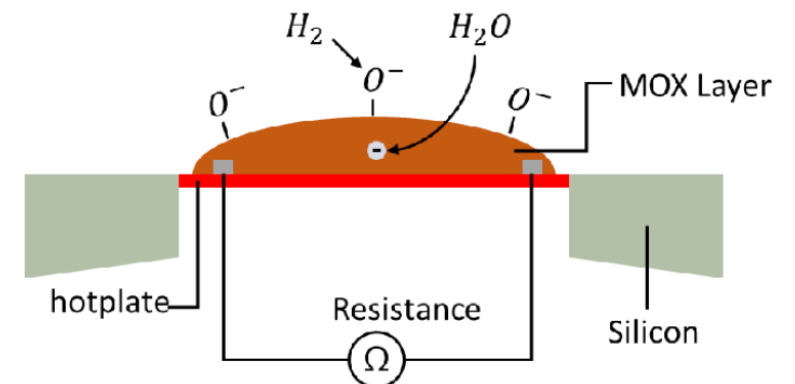
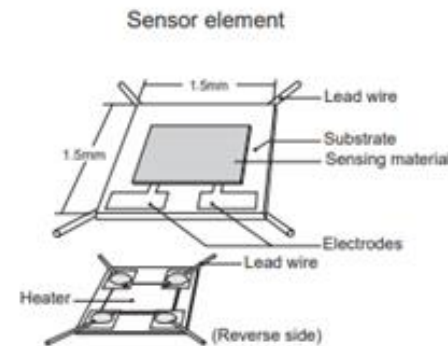
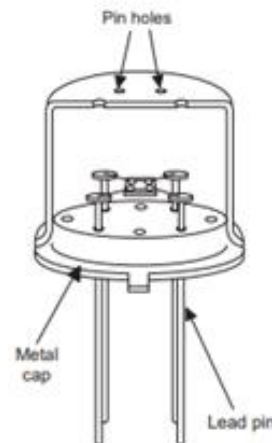
**! MADE BY CSIRO !**

↑ H<sub>2</sub> = ↑ Electrical resistivity

↓ H<sub>2</sub> = ↓ Electrical resistivity



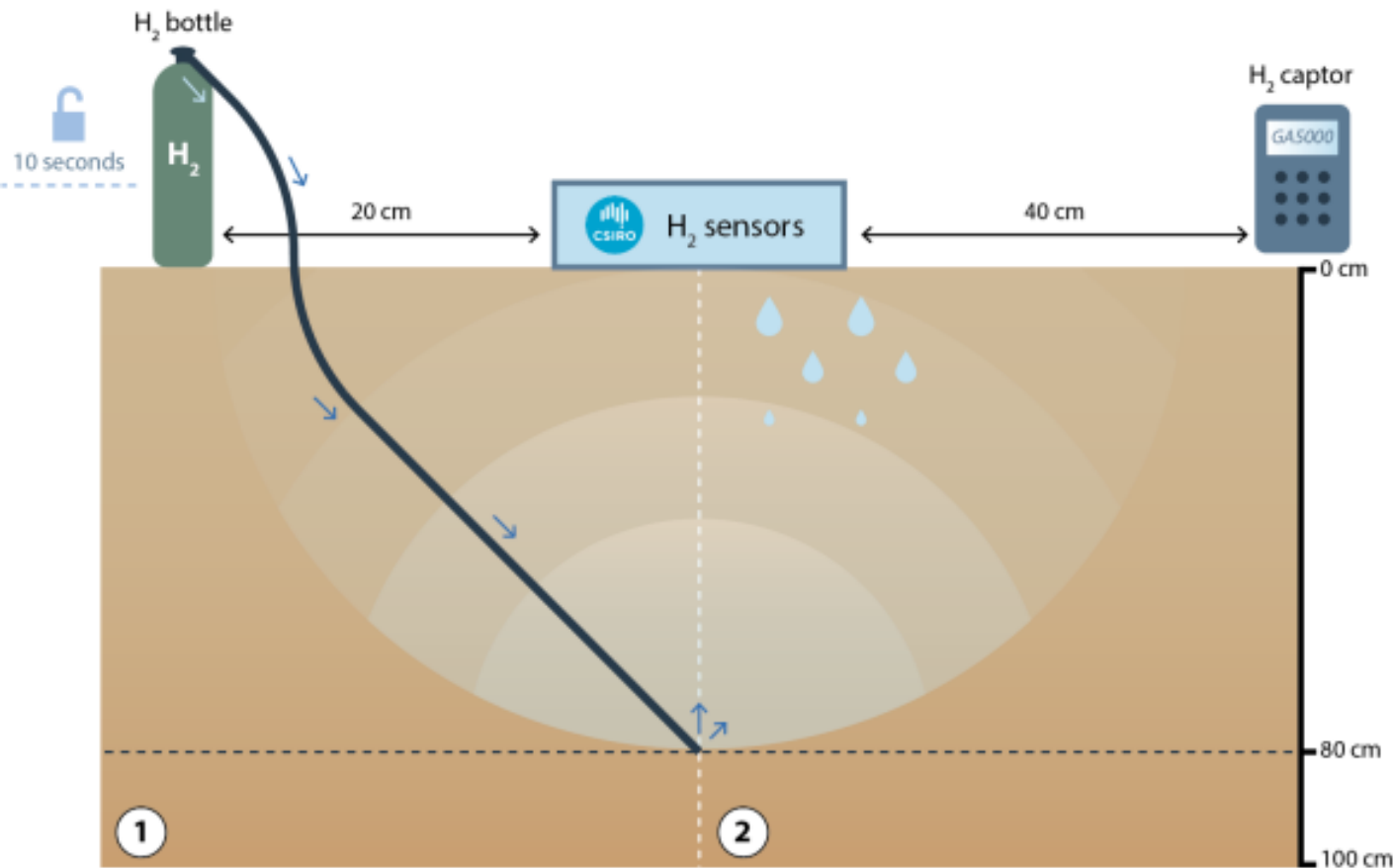
*Sensors and Transmitters. (n.d.). A brief introduction to MOS sensors and their applications*



# Protocol for H<sub>2</sub> injections

- **2<sup>nd</sup> step:** Create (“valuable”) protocol

Protocol sketch



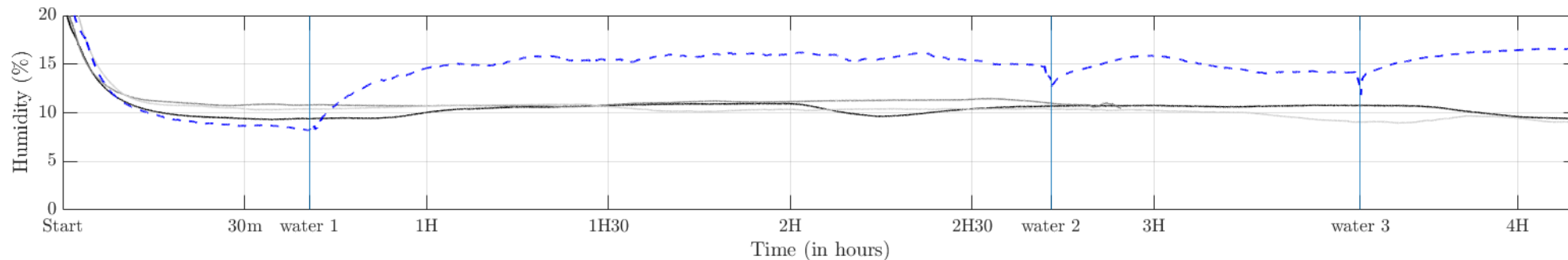
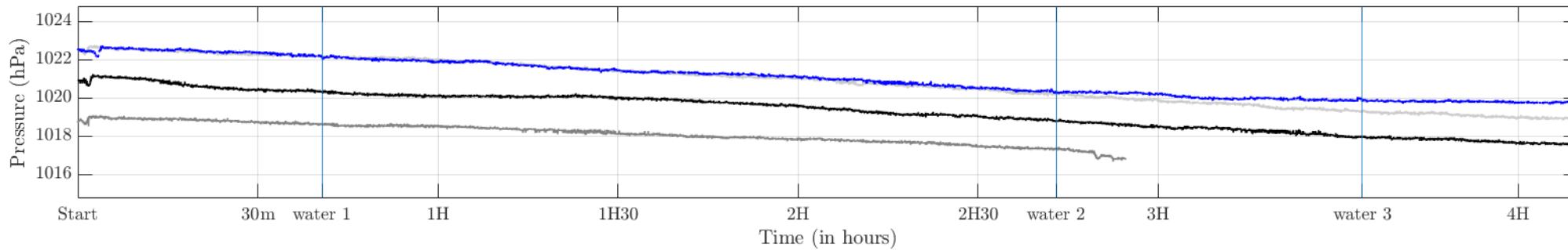
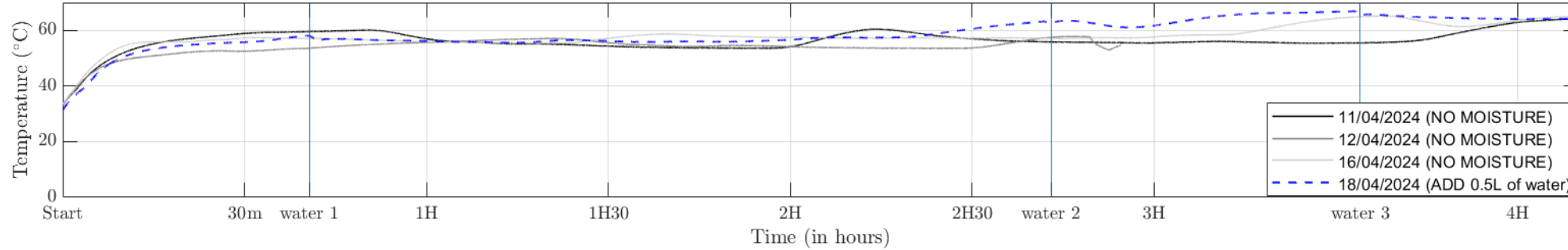
Picture





# Before H<sub>2</sub> injections

- **1<sup>st</sup> step:** Check environmental parameters

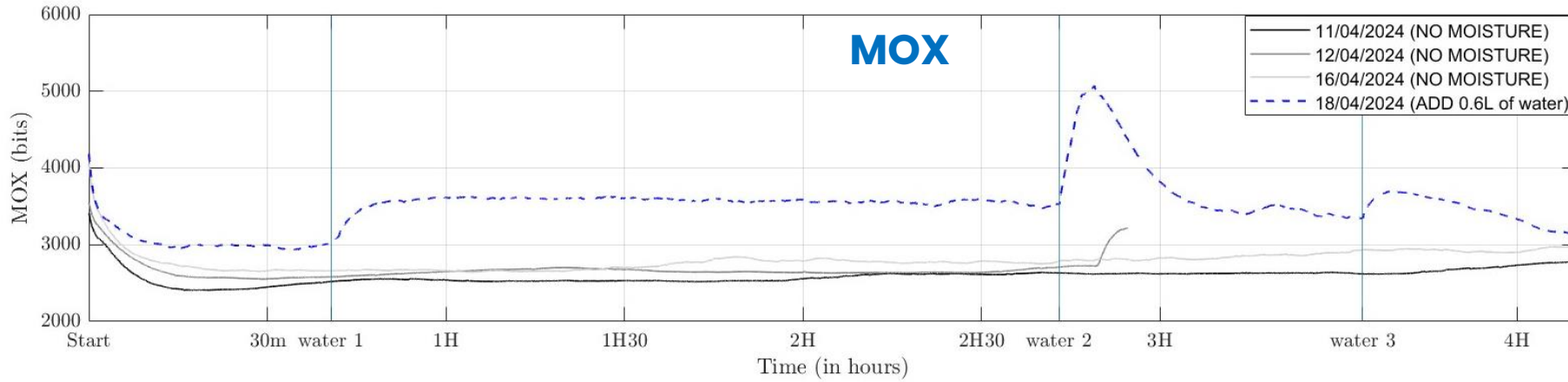


**Essential for :**

- Impact of water injection on each parameter
- Disfunctions

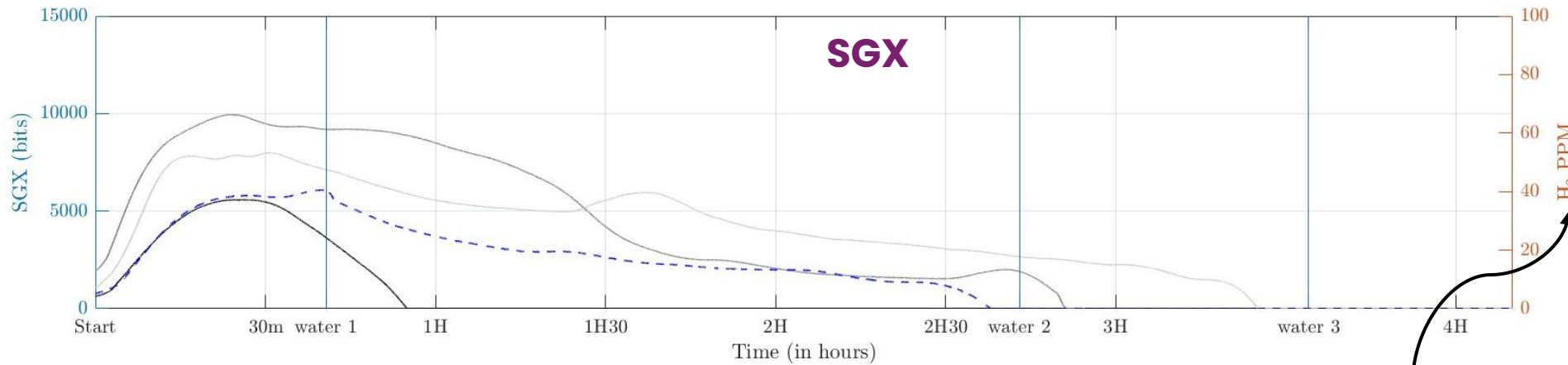
# Before H<sub>2</sub> injections

- **2<sup>nd</sup> step:** Check H<sub>2</sub> sensors



## Essential for :

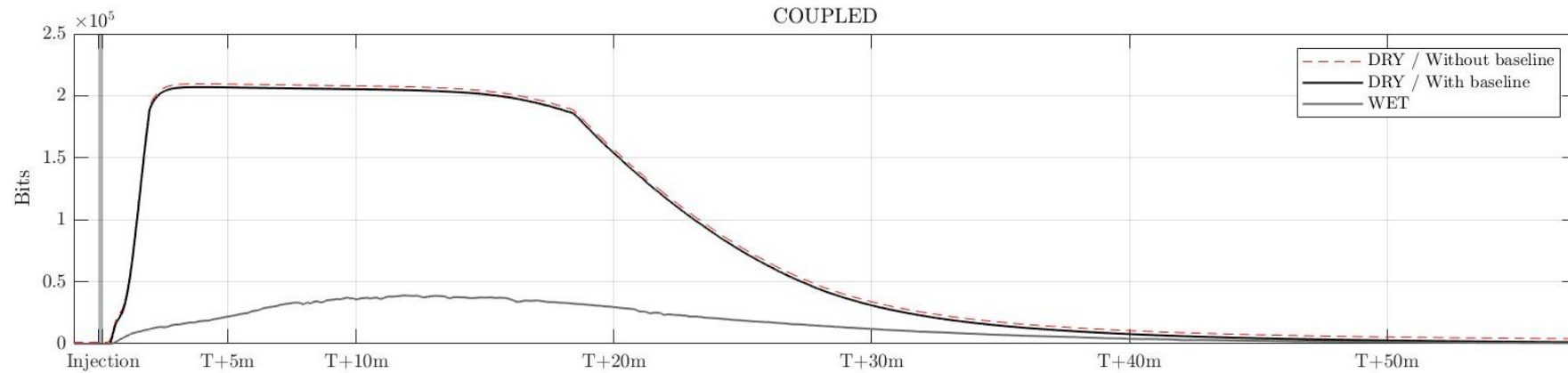
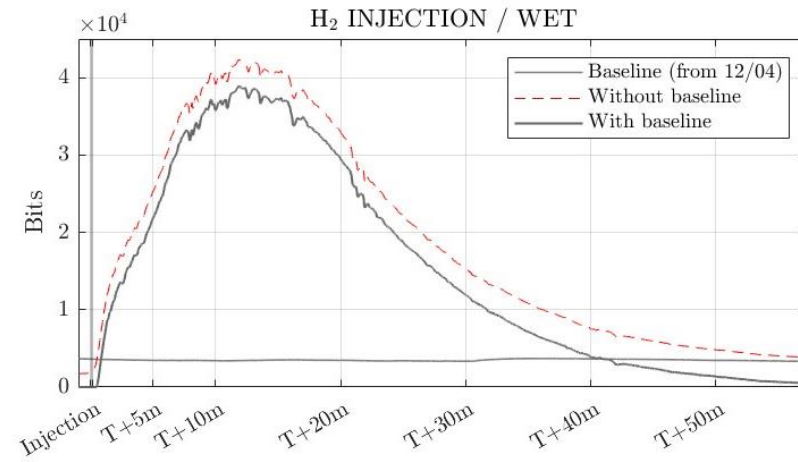
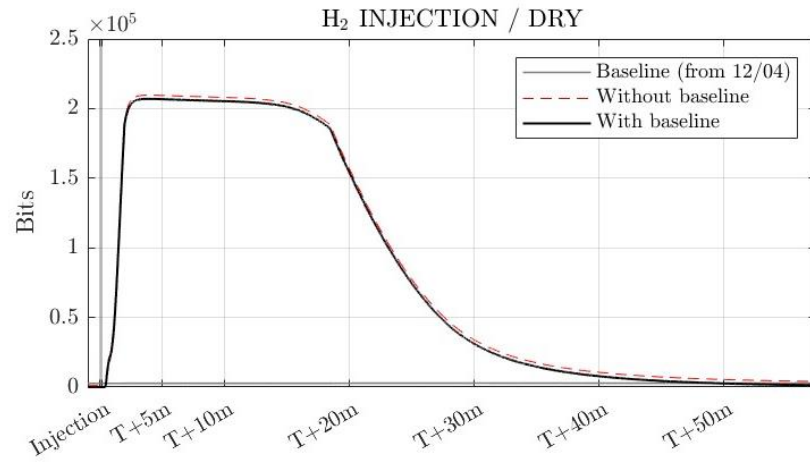
- Sensor time calibration
- Sensor baseline
- Impact of water injection



$$H_2 \text{ (ppm)} = 6.51 * \text{Sensor SGX (bits)} / 1000$$

# H<sub>2</sub> injections

## MOX



## Results

### Sensor

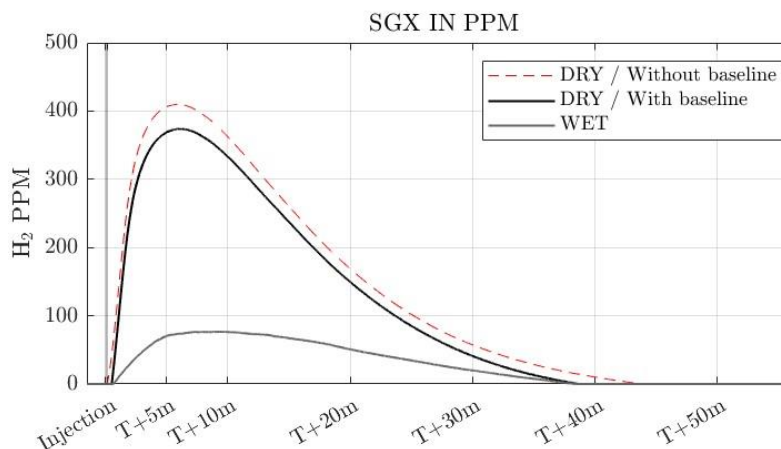
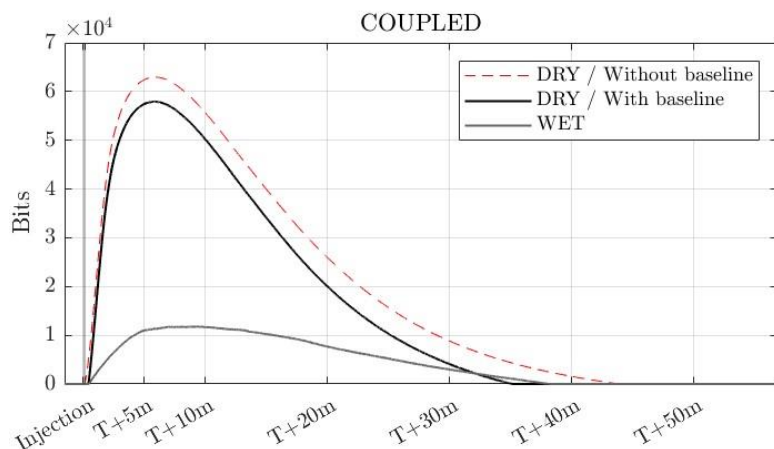
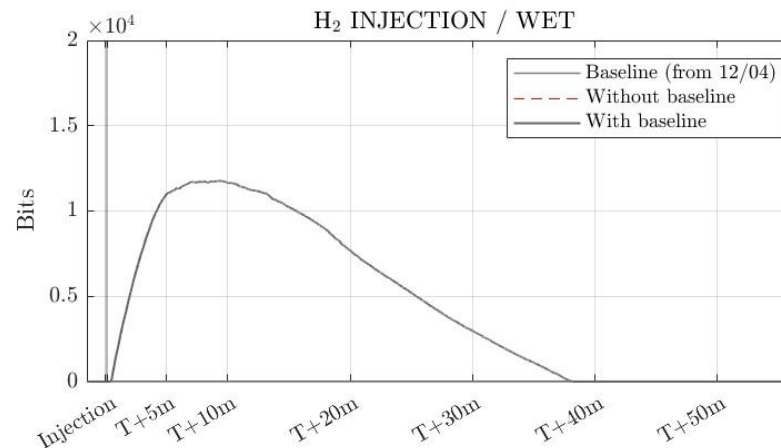
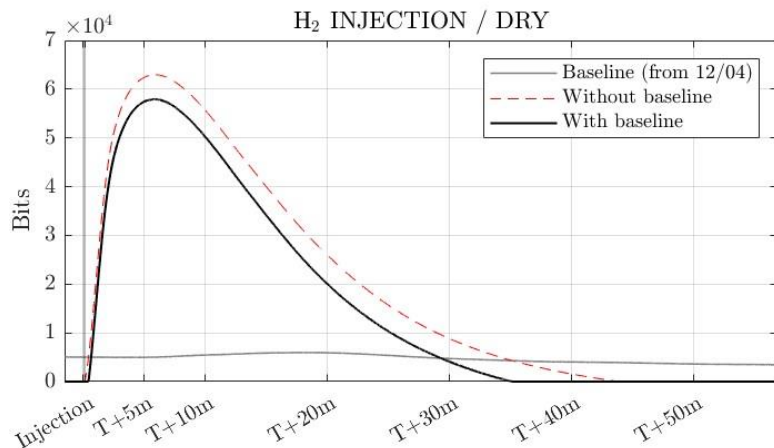
- **Fast response**
- **Sensor saturated for 20 min.**
- **Peak variabilites**

### H<sub>2</sub>

- No H<sub>2</sub> recorded after 50 min.
- High impact on H<sub>2</sub> with wet soil

# H<sub>2</sub> injections

## SGX



### Sensor

- **Fast response**
- **Not saturated**
- **No peak deformation**

### Results

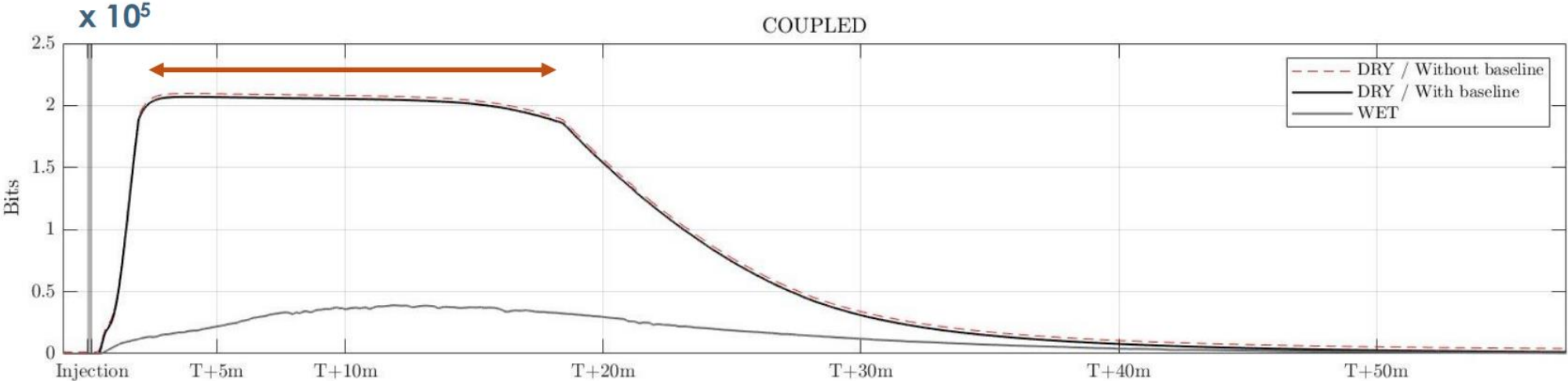
### H<sub>2</sub>

- No H<sub>2</sub> recorded after 40 min.
- High impact on H<sub>2</sub> with wet soil

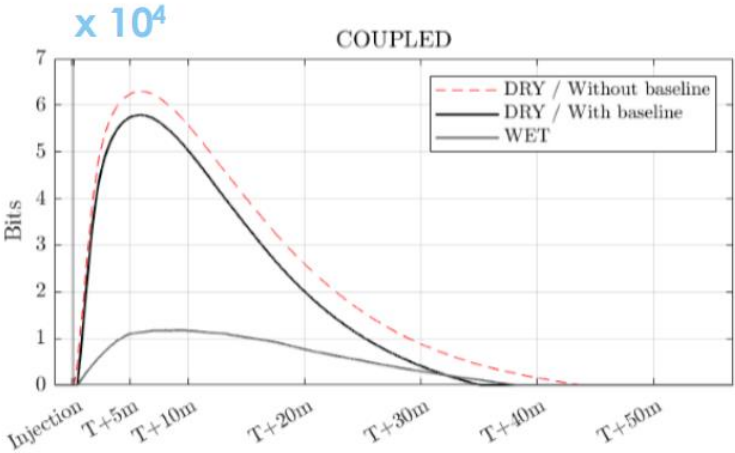
# Conclusions and discussions

- (1) -

MOX



SGX



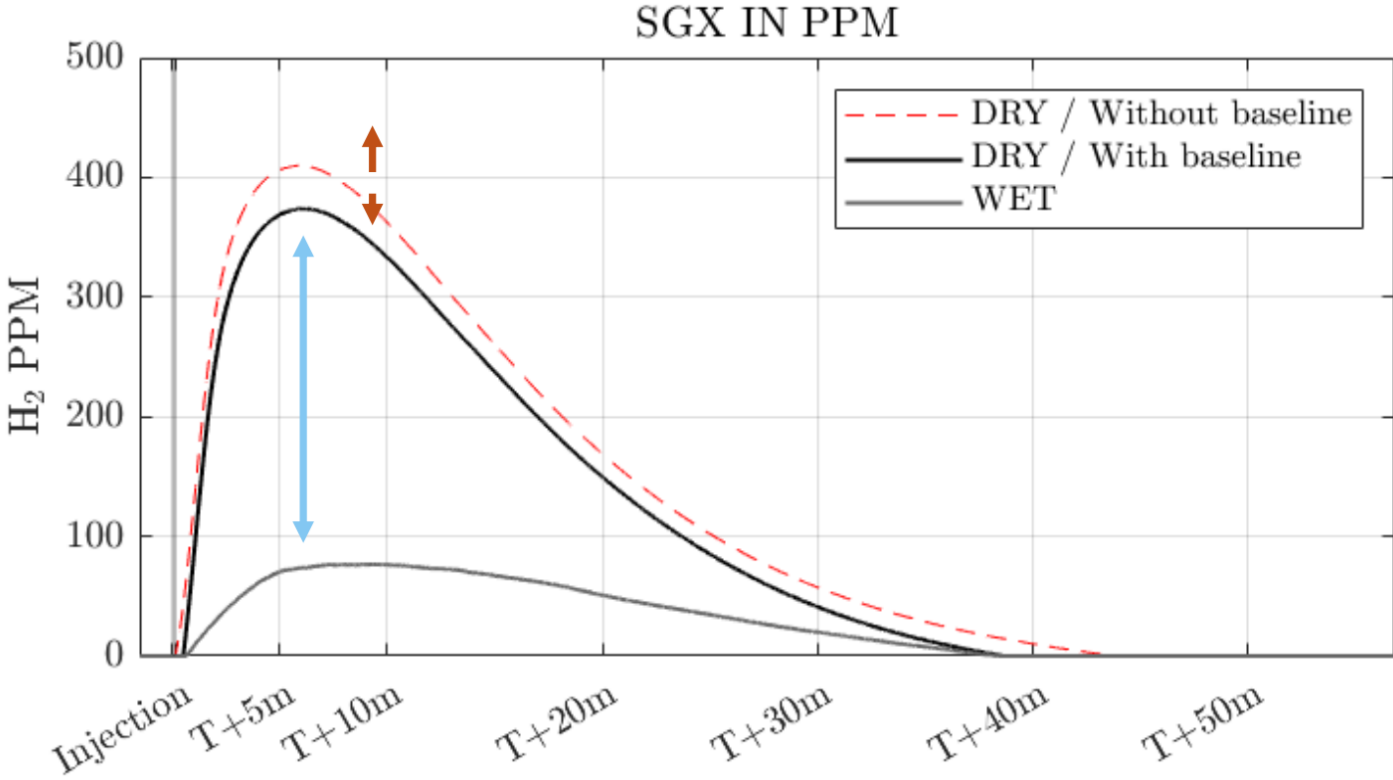
- **MOX** is **saturated** for 15 minutes !




- **MOX** =  $2 \times 10^5$  bits and **SGX** =  $6.5 \times 10^4$  bits

# Conclusions and discussions

- (2) -

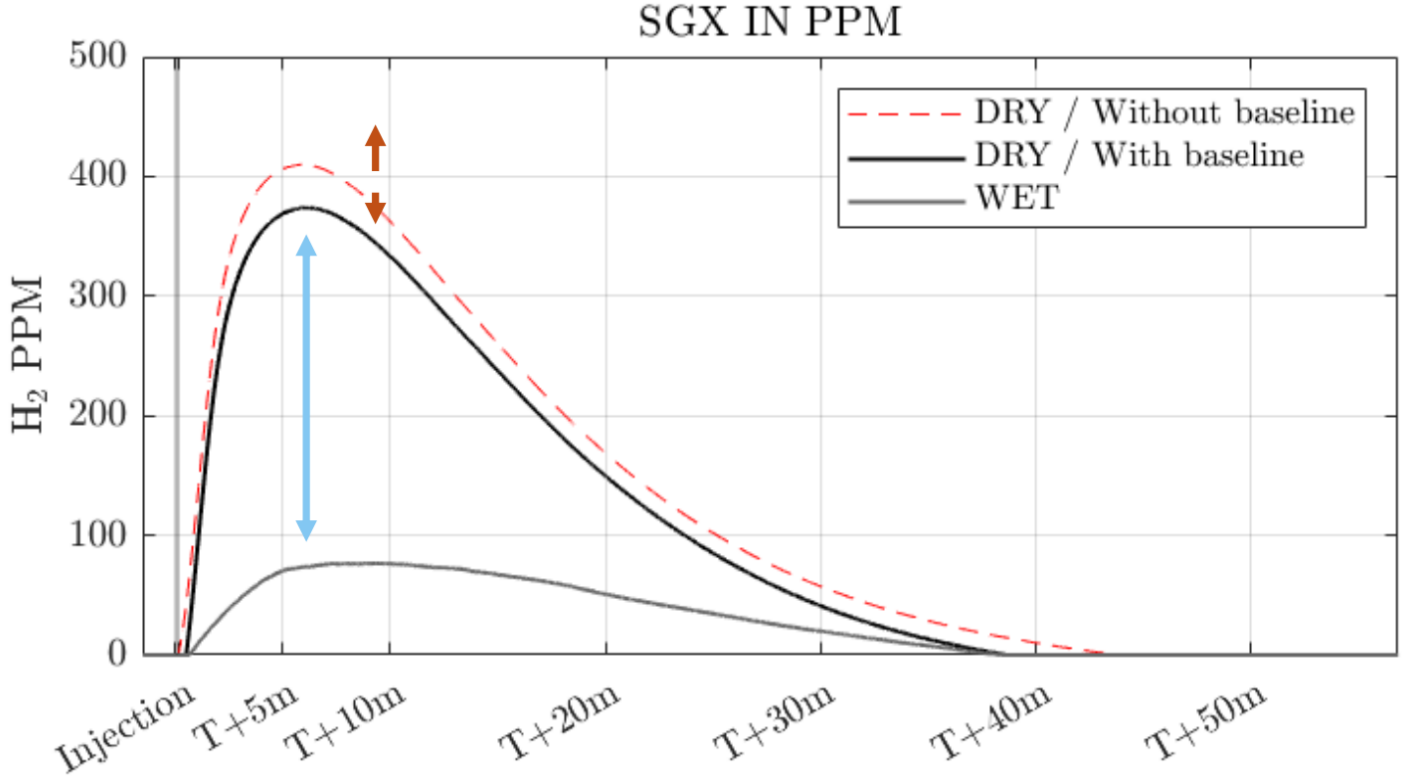


**(a) False positive ... or negative**

 • Anomaly of H<sub>2</sub> up to 65 ppm without baseline (DRY)

# Conclusions and discussions

- (3) -



**(b) H<sub>2</sub> concentration received**

• High variability between dry and wet soil (80% not recorded)

### Why?

- H<sub>2</sub> dissolved in the water ?
- New fluid pathways ?
- Microbial life ?

**Thank you for listening!**

Raphaël Josse